## New Alkylidenecyclopropane Amino Acid **Derivatives for an Efficient Construction of** the 6H-Pyrrolo[3,4-b]pyridine Skeleton

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## Received June 2, 1997

Alkylidenecyclopropanes form a peculiar class of strained olefinic compounds, with remarkable synthetic potential;<sup>1,2</sup> thus, they undergo ring opening with palladium dichloride to produce  $\pi$ -allylpalladium complexes,<sup>3</sup> carbopalladation with vinyl and aryl halides in the presence of Pd(0),<sup>4</sup> regioselective Pd(0)-catalyzed [3 + 2] cycloaddition with olefinic and acetylenic substrates,<sup>1,5</sup> and Pauson-Khand cyclization with dicobalt hexacarbonyl complexes of acetylene.<sup>6,7</sup> Most of these reactions have been reported to occur both inter- and intramolecularly.<sup>4,8</sup> Moreover, alkylidenecyclopropanes constitute the most suitable precursors for cyclobutanone synthesis.<sup>9</sup> As in the case for many cyclopropane derivatives, they are also endowed with specific bioactivities.<sup>10</sup> Optically active alkylidenecyclopropanes have recently been prepared by the regio- and stereoselective Pd(0)catalyzed reduction of asymmetric 1-(1-alkenyl)cyclopropyl esters by sodium formate.<sup>11</sup>

The 6*H*-pyrrolo[3,4-*b*]pyridine ring system **1** (Figure  $1)^{12}$  is of current interest due to its presence in the skeleton of very important antitumor agents such as camptothecin 2 and to the biological properties of new related compounds (e.g., antiretroviral activity, modulation of protein synthesis, ...).<sup>13</sup>

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Figure 1. 6*H*-Pyrrolo[3,4-*b*]pyridine (1) and camptothecin (2).



<sup>a</sup> Key: (a) 6 mol % Pd(dba)<sub>2</sub>, 7.2 mol % dppe, THF, rt; (b) 1 equiv of 4a-c, 1 equiv of NaH, THF, rt, 1 h, 70-95%.

We report here a new and efficient synthesis of optically pure diazabicycloheterocycles derived from 1. Thus, Pd(0)-catalyzed nucleophilic substitution of the 1-vinyl-1-(tosyloxy)cyclopropane  $3^{14}$  (readily available from vinylation and tosylation<sup>15</sup> of cyclopropanone hemiacetal)<sup>16</sup> by the methyl *N*-tosylglycinate (4a) (R = H), (*S*)-(+)-alaninate (4b) ( $\mathbf{R} = \mathbf{Me}$ ), and (*S*)-(-)-phenylalaninate (4c)  $(\mathbf{R} = \mathbf{PhCH}_2)$  in the presence of 1 equiv of NaH produced the methyl N-(2-cyclopropylideneethyl)-N-tosylamino acid esters **5a**, (S)-**5b**, and (S)-**5c** in 81, 95, and 77% yields, respectively; no significant epimerization of the chiral center was observed for (S)-**5b**,**c** (ee >98%) (Scheme 1).<sup>17</sup> It must be noted that the Pd(0)-catalyzed reaction of tosylate 3 with simple amines or imines led exclusively to 2-cyclopropylideneethylamine derivatives;<sup>18</sup> use of N-(diphenylmethylene)glycine esters as the nucleophile, following the same process, was shown to provide  $\alpha$ -allyl- $\alpha$ -amino acids resulting from C-allylation.19

Partial reduction of ester 5a by 0.9 equiv of DIBAH provided the 2-aminoethanal 6a, which was treated directly with methylhydroxylamine hydrochloride in ether in the presence of pyridine (1.1 equiv).<sup>20</sup> The resulting (Z)-nitrone  $7a^{20}$  was not isolated but underwent ready intramolecular 1,3-dipolar cycloaddition<sup>8</sup> to produce exclusively the spiro bicyclic isoxazolidine 8a, in 70% overall yield from **5a**. Otherwise, (S)-**5b**,c were reduced by 2.5 equiv of DIBAH, and the corresponding alcohols were then oxidized under Swern conditions<sup>21</sup> to

(17) Amino acid esters (S)-5b,c were obtained with 98% enantiomeric excesses when 1 equiv of NaH was used, but use of more than 1 equiv of base caused their racemization.

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<sup>a</sup>Key: (a) 0.9 equiv of DIBAH,  $CH_2Cl_2$ , -78 °C, 1 h or (i) 2.5 equiv of DIBAH, (ii) (COCl)<sub>2</sub>, DMSO, iPr<sub>2</sub>EtN; (b) 1.2 equiv of MeNHOH-HCl, 1.4 equiv of pyridine, H<sub>2</sub>O, rt, 18 h, 70-76% overall yield from **5a**-c.

Scheme 3. Transition States of the 1,3-Dipolar Cycloaddition of (Z)-Nitrones 7a-c



provide the 2-aminoethanal (*S*)-**6b**,**c**. Upon treatment with MeNHOH–HCl, (*S*)-**6b** gave the (*Z*)-nitrone (*S*)-**7b**,<sup>20</sup> which underwent cycloaddition to give a 55/45 diastereomeric mixture of the fused cycloadducts (3'a*R*,6'*S*,6'a*R*)-**8b** and (3'a*S*,6'*S*,6'a*S*)-**8'b** in 76% overall yield from the amino ester (*S*)-**5b**, while the aldehyde (*S*)-**6c** (R = PhCH<sub>2</sub>) led after reaction with MeNHOH–HCl via the nitrone (*Z*)-**7c**<sup>20</sup> to a 62/38 diastereomeric mixture of fused cycloadducts (3'a*R*,6'*S*,6'a*R*)-**8c** and (3'a*S*, 6'*S*,6'a*S*)-**8'c** in 70% overall yield (Scheme 2).

The regioselectivity of the cycloaddition, i.e., the exclusive formation of the fused adducts 8a-c and the lack of bridged adducts 9a-c,<sup>22</sup> is supported by simple molecular mechanics calculations (MAD).<sup>23</sup> Thus, the difference of total steric energies between the optimized geometries of the conformations of the transition states leading to cycloadducts 8a-c or 9a-c ( $\Delta E = -15$  kcal/ mol) suggests kinetic control of the reaction. The formation of the two diastereomeric cycloadducts 8b and 8'b observed from the (Z)-nitrone (S)-7b ( $\mathbf{R} = \mathbf{Me}$ ) and of the two diastereomeric cycloadducts 8c and 8'c from the (Z)nitrone (S)-7c ( $R = PhCH_2$ ) probably results from the two possible approaches of the dipolarophile by the (Z)nitrone moieties, either from above (transition state 7A) or from below (transition state 7A') the plane of the alkylidenecyclopropane (Scheme 3). Analogous MAD calculations of the difference of energies between the conformations of the transition states leading to cycloadducts **8b** and **8'b** ( $\Delta E = -1.5$  kcal/mol) or to **8c** and **8'c** ( $\Delta E = -3.6$  kcal/mol), respectively, confirm that the presence of a methyl or benzyl substituent on carbon 6'-C did not induce in these cases any significant diastereo-

Scheme 4<sup>a</sup>



<sup>*a*</sup> Key: (a) xylene reflux, 6 h, 43–64%.

selectivity ascribed to a steric effect.<sup>24</sup> Fortunately, cycloadducts **8b** and **8'b** as well as **8c** and **8'c** were separable by flash chromatography, and their structures were determined by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy. For instance, after irradiation of the methyl on carbon 6'-C at  $\delta$  1.35 ppm, the major isomer **8b** showed a doublet at  $\delta$  3.65 ppm (J = 7.40 Hz) for the 6'-H proton, while the minor isomer **8'b** showed a doublet at  $\delta$  3.20 ppm (J = 5.36 Hz) for the same proton; as  $J_{cis}$  is larger than  $J_{trans}$  in isoxazolidine and pyrrolidine rings,<sup>22</sup> the observed coupling constants supported the structure assignments.

On heating in xylene at reflux for 6 h, the N–O isoxazolidine bond of the tricyclic isoxazolidines **8a**–c was cleft to produce the cyclopropyloxy diradicals **10a**–c, which then readily underwent cyclopropane ring opening with strain release into the diradicals **11a**–c, followed by ring reclosure, to afford the diazaheterocycles **12a** (49%) and (1*R*,6*S*,9*S*)-**12b**,c (43–64%, Scheme 4). The isoxazolidines **8'b**,c underwent the thermal induced ring expansion to give the diastereomeric (1*S*,6*R*,9*S*)-diazaheterocycles. Epimerization of diamines **12b**,c was not observed under these conditions<sup>25</sup> and the whole process occurred with the maximum "atom economy" as no added reagent or catalyst was required after nitrone formation.

Now readily available, the derivatives of 2,4diazabicyclo[3.4]octan-7-ones **12a**–**c** are specific substance P (SP) antagonists both in vitro and in vivo (IC<sub>50</sub> ranges from 0.8 to 3  $\mu$ M for the NK<sub>1</sub> receptor of human IM<sub>9</sub> cells<sup>26,27</sup>); they appeared as promising therapeutic agents for the treatment of important diseases (asthma, inflammation and pain, migraine, and vascular headaches).<sup>28</sup> Moreover, they constitute suitable precursors of bioactive alkaloids such as camptothecin (**2**), for instance, and analogous derivatives. Synthetic applications of these new diazaheterocycles are currently under investigation.

Acknowledgment. This work was financially supported by the CNRS (France) and the CNR (Italy) within a French–Italian Cooperation Program and through grants from the ERASMUS European Program. We are grateful to Dr. Jean-François Peyronel (Rhone-Poulenc Rohrer, France) for helpful discussions concerning the SP antagonists.

**Supporting Information Available:** Full experimental procedures and complete spectral data for all reaction products (8 pages).

JO9709615

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<sup>(23)</sup> Molecular mechanics calculations have been performed with the MAD (Molecular Advanced Design) version 2.3, using a MM2-type force field, available from Oxford Molecular, Ecole Polytechnique, Palaiseau (France). The minima were found through a complete geometrical optimization of the structure for which the distances between the atoms  $C_{3'}-O_{2'}$  and  $C_{2'a}-C_{6'a}$  were constrained to 3 Å.

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